

# CH 003: INTRODUCTORY GENERAL CHEMISTRY

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**Originator**

cmelton

**Justification / Rationale**

To offer our students more online options for the safety of the students, faculty, and staff during the pandemic, but also to accommodate more of our students who may need access to online courses in order to attend.

**Effective Term**

Fall 2022

**Credit Status**

Credit - Degree Applicable

**Subject**

CH - Chemistry

**Course Number**

003

**Full Course Title**

Introductory General Chemistry

**Short Title**

INTRO GENERAL CHEM

**Discipline****Disciplines List**

Chemistry

**Modality**

Face-to-Face

Hybrid

**Catalog Description**

This course is the study of the principles of inorganic chemistry. Instruction is given in calculations, atomic theory, periodic law, compounds, bonding, nomenclature, states of matter, reaction types, composition, stoichiometry, acids, bases, pH, gas laws, and solutions. Laboratory is a hands-on experience stressing manipulation, procedure, data, and outcome calculations. Note: This course should be taken by Liberal Studies majors on the Education track to satisfy the chemistry requirement. It should also be taken as a prerequisite for CH 001A for students who have not had high school chemistry within the past 5 years.

**Schedule Description**

This course gives an overview of inorganic chemistry and teaches the students the proper chemical laboratory techniques.

Prerequisite: MATH 054 Advisory: ENG 061 IGETC: 5A\*, 5C\*

**Lecture Units**

3

**Lecture Semester Hours**

54

**Lab Units**

1

**Lab Semester Hours**

54

**In-class Hours**

108

**Out-of-class Hours**

108

**Total Course Units**

4

**Total Semester Hours**

216

**Prerequisite Course(s)**

MATH 054

Advisory: ENG 061

**Required Text and Other Instructional Materials****Resource Type**

Book

**Author**

Zumdahl-Decoste

**Title**

Introductory Chemistry: A Foundation

**Edition**

9th

**Publisher**

Cengage Learning

**Year**

2014

**College Level**

Yes

**Flesch-Kincaid Level**

12

**ISBN #**

978-1337399425

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**Resource Type**

Manual

**Author**

Hall, J., Little, J.

**Title**

Introductory Chemistry in the Laboratory

**Publisher**

Cengage Learning

**Year**

2015-01-01

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**For Text greater than five years old, list rationale:**

This text has not been updated, better materials have not been published.

**Class Size Maximum**

24

**Entrance Skills**

Explain the real number system: integers, rational and irrational numbers.

**Requisite Course Objectives**

MATH 054-Identify, recognize and classify real numbers, as integers, rationals, or irrationals and locate their approximate positions on the real number line.

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**Entrance Skills**

Understand the concept of a variable.

**Requisite Course Objectives**

MATH 054-Understand the concepts of variables and how variables can be used to represent an unknown quantity or a range of quantities.

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**Entrance Skills**

Use variables to generate algebraic expressions modeling an application (word) problem.

**Requisite Course Objectives**

MATH 054-Use variables to create algebraic expressions that model quantities in an application problem.

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**Entrance Skills**

Demonstrate arithmetic of algebraic expressions, including the use of the commutative, associative, distributive, identity, and inverse properties, the use of the order of operations, and the use of integer exponents and the rules of exponents.

**Requisite Course Objectives**

MATH 054-Use the properties of integer exponents to simplify algebraic expressions, including expressions involving scientific notation.

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**Entrance Skills**

Create equations that model real world situations given in application (word) problems.

**Requisite Course Objectives**

MATH 054-Use variables to create algebraic expressions that model quantities in an application problem.

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**Entrance Skills**

Advisory skill: Demonstrate a critical thinking skills when reading, composing and participating in class discussions.

**Requisite Course Objectives**

ENG 061-Demonstrate the ability to think critically and express ideas using various patterns of development.

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**Entrance Skills**

Advisory skill: Demonstrate the ability to read and respond in writing beyond the literal interpretation of the text.

**Requisite Course Objectives**

ENG 061-Demonstrate the ability to read and respond in writing beyond the literal interpretation of the text.

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**Entrance Skills**

Advisory skill: Develop, organize and express complex ideas in both expository and research papers.

**Requisite Course Objectives**

ENG 061-Use theses to organize paragraphs into coherent analyses.

ENG 061-Demonstrate the ability to use research skills including library resources such as books, periodicals, electronic databases and online resources such as the internet.

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**Course Content**

1. Chemistry: An Introduction
2. Measurements and Calculations
3. Matter
4. Chemical Foundations: Elements, Atoms and Ions
5. Nomenclature
6. Chemical Reactions: An Introduction
7. Reactions in Aqueous Solutions
8. Chemical Composition
9. Chemical Quantities
10. Energy
11. Modern Atomic Theory
12. Chemical Bonding
13. Gases
14. Liquids and Solids
15. Solutions
16. Acids and Bases
17. Equilibrium
18. Oxidation-Reduction Reactions and Electrochemistry

**Lab Content**

1. Safety Rules & Laboratory measurements: Balances
2. Use of Volumetric Glassware
3. Density determination of a solid, liquid and of a solution
4. Calorimetry: Calculation of the specific heat of a metal
5. Metathesis (double displacement) reactions (precipitation)
6. Oxidation-reduction reactions
7. Graphing the data
8. Determine the % composition of MgO by reacting Mg with O<sub>2</sub>. Calculate the empirical formula
9. Stoichiometry and Limiting reactant
10. Line Spectra
11. Lewis Structures and Molecular Shapes
12. Molar mass of a volatile liquid
13. Properties and Reactions of acids and bases; measurement of pH
14. Acid-Base Titrations
15. Scientific Abstract

**Course Objectives**

| Objectives  |  |
|-------------|--|
| Objective 1 | Explain the fundamental concepts of chemistry with their applications to human affairs.  |
| Objective 2 | Discuss major historical chemistry discoveries which have been a part of human social and technological development.   |
| Objective 3 | Demonstrate use of tools and instruments involved in making findings in chemical behavior.   |
| Objective 4 | Solve chemistry problems with the coupled recognition that calculation methods in chemistry are shared in other domains such as business, economics, and technology. |
| Objective 5 | Demonstrate the ability to raise questions and how to formulate them clearly   |
| Objective 6 | Demonstrate the ability to collect and interpret the data  |
| Objective 7 | Demonstrate the ability to work in teams and to respect other people's opinions  |

**Student Learning Outcomes**

| Upon satisfactory completion of this course, students will be able to: |   |
|--|---|
| Outcome 1  | Analyze quantitative data to draw plausible conclusions.                            |
| Outcome 2  | Relate the macroscale phenomena of daily life to microscale atomic concepts.        |
| Outcome 3  | Apply scientific literacy to vet sources and summarize observed chemical phenomena. |
| Outcome 4  | Perform basic general chemistry laboratory experiments safely and accurately.       |

**Methods of Instruction**

| Method        | Please provide a description or examples of how each instructional method will be used in this course.   |
|---------------|--|
| Journal       | Supplemental readings focusing on historical events in chemistry and the every-day applications of chemistry develops an awareness and understanding of chemistry beyond the concise principles offered in the assigned text |
| Experiential  | Hands-on labs help students understand how to run experiments and observation of experiments in real time.   |
| Discussion    | Lecture and lab discussions offer students an opportunity to critically think about concepts.  |
| Participation | Student have the opportunity to participate in lecture via discussion and example problems, and in lab via lab experiments.  |
| Observation   | Observing students perform labs with opportunities to correct methods along the way.   |
| Lecture       | Course concepts will be presented to students in a lecture format.   |
| Laboratory    | Students will be able to perform laboratory experiments weekly.  |

**Methods of Evaluation**

| Method                                   | Please provide a description or examples of how each evaluation method will be used in this course.  | Type of Assignment  |
|--|--|---------------------|
| Guided/unguided journals                 | A typed scientific abstract based on an assigned reading of a chemical article.  | Out of Class Only   |
| Written homework                         | Homework sets with approximately 30 problems per week will be assigned in each chapter with answers in the back of the text for self check.  | Out of Class Only   |
| Mid-term and final evaluations           | A comprehensive final examination will be administered covering all previously completed topics for the semester. Questions will require problem solving, short answer and matching.   | In Class Only       |
| Tests/Quizzes/Examinations               | An examination will be given covering each topic area in that section. The in class examinations will consist of statement answers, multiple choice, fill in the blank, and problem solving. There will be 4-8 exams, 15-17 quizzes. | In and Out of Class |
| Group activity participation/observation | In the laboratory students will have the opportunity to work in groups and observe chemical demonstrations.  | In Class Only       |
| Laboratory projects                      | In the lab, students will perform chemical experiments for data collection and analysis.   | In Class Only       |

**Assignments**
**Other In-class Assignments**

Quizzes and exams  
Lab experiments

### Other Out-of-class Assignments

1. Textbook reading, approximately 25 pages per week.
2. Textbook homework, approximately 25-30 problems per week.
3. Complete and receive certification of completion for all pre-laboratory assignment sheets. Approximately 15 assignments.
4. Complete and receive certification of completion for all pre-laboratory procedure write-ups. Approximately 15 assignments.
5. Submit laboratory work for evaluation and completeness and correctness of data and calculations. Approximately 15 lab experiments (listed under course content)

### Grade Methods

Letter Grade Only

### Distance Education Checklist

Include the percentage of online and on-campus instruction you anticipate.

Online %

50

On-campus %

50

### Lab Courses

**How will the lab component of your course be differentiated from the lecture component of the course?**

Labs will be face-to-face and focused on performing experiments. Lectures will be focused on the theory and problem-solving aspects of chemistry.

**From the COR list, what activities are specified as lab, and how will those be monitored by the instructor?**

Face-to-face delivery of chemical laboratory experiments.

**How will you assess the online delivery of lab activities?**

n/a

### Instructional Materials and Resources

**If you use any other technologies in addition to the college LMS, what other technologies will you use and how are you ensuring student data security?**

Zoom to host synchronous meetings and office hours. Security will be ensured by the use of a course pass-code.

**If used, explain how specific materials and resources outside the LMS will be used to enhance student learning.**

Email, zoom, and phone will allow students more options for contact with faculty as well as live help or Q&A.

### Effective Student/Faculty Contact

**Which of the following methods of regular, timely, and effective student/faculty contact will be used in this course?**

**Within Course Management System:**

Chat room/instant messaging  
Discussion forums with substantive instructor participation  
Online quizzes and examinations  
Private messages  
Regular virtual office hours  
Timely feedback and return of student work as specified in the syllabus  
Video or audio feedback  
Weekly announcements

**External to Course Management System:**

Direct e-mail  
Posted audio/video (including YouTube, 3cm mediasolutions, etc.)  
Synchronous audio/video  
Teleconferencing  
Telephone contact/voicemail

**For hybrid courses:**

Orientation, study, and/or review sessions

**Briefly discuss how the selected strategies above will be used to maintain Regular Effective Contact in the course.**

Synchronous Zoom time for class and office hours will allow direct conversation with students. On top of that, regular weekly announcements will update students about course happenings, such as upcoming due dates or recently graded assignments. Email, chat, discussion boards, zoom office hours, and phone calls/voicemail will enable students to keep in regular touch if they have questions and for the professor to check in with them regularly as needed. Posted audio/video can be used to deliver lectures or course messages. Last, assignment feedback and rubrics will give students a good idea of how to do assignments and what mistakes might be occurring, so they can make corrections.

**If interacting with students outside the LMS, explain how additional interactions with students outside the LMS will enhance student learning.**

Email, zoom, and phone will allow students more options for contact with faculty as well as live help or Q&A.

**Other Information****Provide any other relevant information that will help the Curriculum Committee assess the viability of offering this course in an online or hybrid modality.**

All posted videos have been closed captioned through 3CMedia.

**COD GE**

C1 - Natural Sciences

**CSU GE**

B1 - Physical Science

B3 - Laboratory Activity

**IGETC GE**

5A - Physical Science

5C - Science Laboratory

**MIS Course Data****CIP Code**

40.0501 - Chemistry, General.

**TOP Code**

190500 - Chemistry, General

**SAM Code**

E - Non-Occupational

**Basic Skills Status**

Not Basic Skills

**Prior College Level**

Not applicable

**Cooperative Work Experience**

Not a Coop Course

**Course Classification Status**

Credit Course

**Approved Special Class**

Not special class

**Noncredit Category**

Not Applicable, Credit Course

**Funding Agency Category**

Not Applicable

**Program Status**

Program Applicable

**Transfer Status**

Transfer CSU, limited UC

**General Education Status**

Y = Not applicable

**Support Course Status**

N = Course is not a support course

**C-ID**

CHEM 101

**Allow Audit**

No

**Repeatability**

No

**Materials Fee**

No

**Additional Fees?**

No

**Approvals****Curriculum Committee Approval Date**

11/02/2021

**Academic Senate Approval Date**

11/11/2021

**Board of Trustees Approval Date**

12/17/2021

**Chancellor's Office Approval Date**

02/03/2022

**Course Control Number**

CCC000333773

**Programs referencing this course**Liberal Arts: Math and Science AA Degree (<http://catalog.collegeofthedesert.eduundefined/?key=29>)Agri-Business AS Degree (<http://catalog.collegeofthedesert.eduundefined/?key=46>)General Agriculture AS Degree (<http://catalog.collegeofthedesert.eduundefined/?key=49>)Elementary Teacher Education AA-T Degree (<http://catalog.collegeofthedesert.eduundefined/?key=5>)Nutrition and Dietetics AS-T Degree (<http://catalog.collegeofthedesert.eduundefined/?key=7>)Agriculture Food Safety Certificate of Achievement (<http://catalog.collegeofthedesert.eduundefined/?key=83>)Agriculture Office Assistant Certificate of Achievement (<http://catalog.collegeofthedesert.eduundefined/?key=84>)Agriculture Office Professional Certificate of Achievement (<http://catalog.collegeofthedesert.eduundefined/?key=85>)Agriculture Pest Management Certificate of Achievement (<http://catalog.collegeofthedesert.eduundefined/?key=86>)Agriculture Technician Certificate of Achievement (<http://catalog.collegeofthedesert.eduundefined/?key=87>)Agriculture Plant Science AS-T Degree (<http://catalog.collegeofthedesert.eduundefined/?key=89>)Agriculture Irrigation Technician Certificate of Achievement (<http://catalog.collegeofthedesert.eduundefined/?key=91>)